



Phase formation in intermixed Ni–Ge thin films: Influence of Ge content and low-temperature nucleation of hexagonal nickel germanides



B. De Schutter^{a,*}, W. Devulder^a, A. Schrauwen^b, K. van Stiphout^b, T. Perkisas^c, S. Bals^c, A. Vantomme^b, C. Detavernier^a

^a Dept. of Solid-State Sciences, Ghent University, Krijgslaan 281/S1, 9000 Ghent, Belgium

^b Instituut voor Kern- en Stralingsfysica, KU Leuven, Celestijnenlaan 200D, B-3001 Leuven, Belgium

^c EMAT, University of Antwerp, Groenenborgerlaan 171, 2020 Antwerp, Belgium

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ABSTRACT

In this study, we focus on phase formation in intermixed Ni–Ge thin films as they represent a simplified model of the small intermixed interface layer that is believed to form upon deposition of Ni on Ge and where initial phase formation happens. A combinatorial sputter deposition technique was used to co-deposit a range of intermixed Ni–Ge thin films with Ge concentrations varying between 0 and 50 at.%Ge in a single deposition on both Ge (100) and inert SiO₂ substrates. *In situ* X-ray diffraction and transmission electron microscopy were used to study phase formation. In almost the entire composition range under investigation, crystalline phases were found to be present in the as-deposited films. Between 36 and 48 at.%Ge, high-temperature hexagonal nickel germanides were found to occur metastably below 300 °C, both on SiO₂ and Ge (100) substrates. For Ge concentrations in the range between 36 and 42 at.%, this hexagonal germanide phase was even found to be present at room temperature in the as-deposited films. The results obtained in this work could provide more insight in the phase sequence of a pure Ni film on Ge.

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1. Introduction

The continuous scaling down of microelectronic devices pushes the current Si-based technology to its limits. Alternative materials are being investigated to replace Si in e.g. the gate regions of Metal-Oxide-Semiconductor Field Effect Transistors (MOSFET). According to the industry roadmap, Ge is a top candidate to replace Si in p-channel MOSFET devices due to its higher carrier mobility and relative compatibility with silicon processing [1]. Similar to the current SALICIDE (Self-Aligned Silicide) process, where metal silicides are used to contact the source, gate and drain regions of MOSFETs, metal germanides appear as natural candidates for contacting Ge. A systematic study of germanide formation and properties in 20 transition metal on Ge systems by Gaudet et al. [2], revealed NiGe as the most promising contact material since it exhibits the most suitable properties among all investigated metal germanides, including low formation temperature, low resistivity and a wide stable temperature window during ramp anneals.

If we are to use NiGe as a contact material in future devices, a thorough understanding of the formation mechanisms and properties of the different germanides in the Ni–Ge system is important.

For the bulk system, most of the work on this binary Ni–Ge system was done in the 1970s [3,4] and was summarized in 1987 by Nash and Nash [5], which led to the Ni–Ge binary phase diagram as it is known today (see Fig. 1). For the thin film system, detailed studies on the phase formation upon annealing of a thin Ni film on Ge using *in situ* X-ray diffraction (XRD) were performed by Gaudet et al. [6] and Nemouchi et al. [7]. Both groups reported the simultaneous growth of NiGe and Ni₅Ge₃ on Ge (100), where Ni₅Ge₃ only exists over a small temperature window since it is consumed by the growth of NiGe, which is the stable end phase. However, the existence of other Ni-germanides in this intermediate region could not be ruled out due to overlapping peak positions of different Ni-rich germanides [6]. Interestingly, Jensen et al. [8] reported on the metastable nucleation of the high temperature ϵ -Ni₅Ge₃ when annealing ratio-controlled Ni–Ge multilayers on an inert SiO₂ substrate.

In order to fully understand the phase formation in the Ni–Ge system, more information on this Ni-rich phase region is required. In the Ni–Si system, a similar region of transient Ni-rich phases during a ramp anneal of a thin Ni film on Si can be observed. Recently, a detailed study of a ratio-controlled Ni–Si system was performed in our group to gain more insight in the early phase formation in the Ni–Si system, which led to the identification of the metastable hexagonal θ -phase in the Ni on Si phase sequence

* Corresponding author. Tel.: +32 9 264 43 70.

E-mail address: bob.deschutter@ugent.be (B. De Schutter).

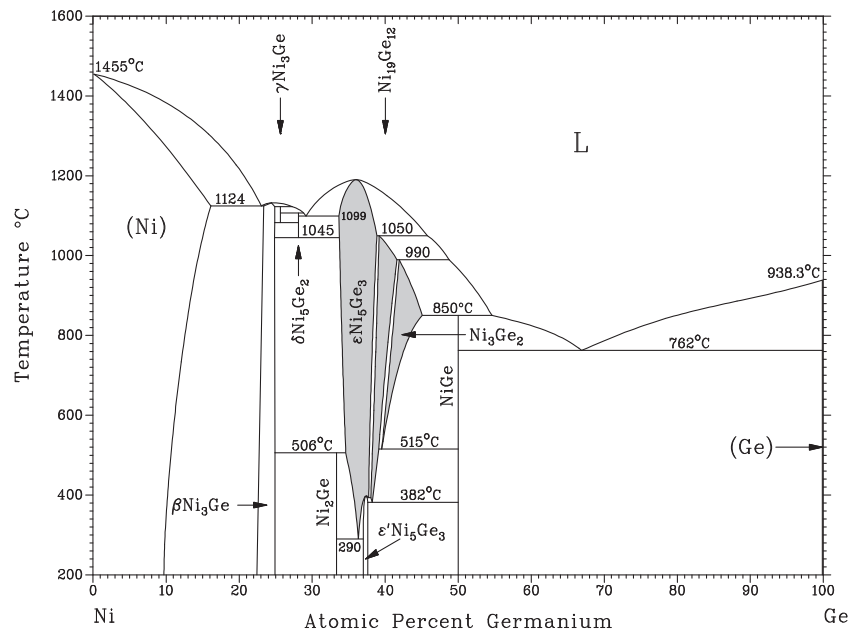


Fig. 1. Equilibrium phase diagram of the Ni–Ge system. This study focuses on the 0–50 at.%Ge composition region. (taken from [5]).

[9–12]. In this paper, we present a similar study of phase formation in the Ni–Ge system. Ratio-controlled Ni–Ge films were deposited on both SiO₂ and Ge (100) substrates and *in situ* XRD was used to monitor phase formation during linear ramp anneals.

2. Experimental

Intermixed Ni–Ge films with varying Ge concentrations (from 0 to 50 at.%Ge) were deposited on both 100 nm thermal SiO₂ and HF-cleaned p-type Ge (100) substrates using a combinatorial sputter deposition technique. Substrate strips with a length of 15 cm and a width of 1–2 cm were mounted on a rotating carousel in a deposition chamber with a base pressure of 2×10^{-7} mbar. During co-sputtering of Ni and Ge, spatial control of the separate material fluxes resulted in a Ge film with a thickness gradient being intermixed with a uniform Ni film. The deposition parameters were tuned in such a way that the Ni content in the resulting film is constant and corresponds to the amount of Ni in a pure 50 nm Ni film (i.e. about 45×10^{16} atoms/cm²). The outcome of such a single deposition is a 15 cm long strip with a Ni–Ge film in which the Ge concentration varies from 0 to 50 at.%. This gradient is then cleaved into individual 5 mm long samples for further characterization. This effectively results in 29 samples per deposited strip with a composition difference of ~ 2 at.% between consecutive samples and a Ge concentration uniformity of ± 1 at.% within a single sample. The composition of all samples was verified with Rutherford Backscattering Spectrometry (RBS).

Germanide phase formation was studied using a home-built *in situ* XRD setup, consisting of an experimental heating chamber mounted in a Bruker D8 Discover XRD system. All individual samples were subjected to a ramp anneal at 3 °C/s from room temperature up to a temperature of 800 °C or 650 °C for the samples on SiO₂ and Ge (100) respectively. During the anneal, the diffraction pattern was recorded every 3 s over a range of 20° in 2 θ .

High Resolution Transmission Electron Microscopy (HR-TEM) was performed on a FEI Tecnai G2 electron microscope, operating at 200 kV. Sample preparation was done using mechanical polishing followed by ion milling.

3. Results

3.1. Ni (Ge) on inert SiO₂ substrates

Due to the inert SiO₂ substrate, the Ni–Ge layers have a fixed Ni/Ge composition throughout the anneal. As can be expected from the phase diagram (Fig. 1), the amount of Ge in the Ni (Ge) mixture will have an influence on the initial crystallization temperature and the first forming phase. Here, *in situ* XRD was used to probe the phase formation sequence while heating the samples to 800 °C.

Fig. 2(a) shows selected *in situ* XRD measurements with 2 θ on the vertical axis and temperature on the horizontal axis. The measured XRD intensity is plotted as a logarithmic grayscale map (with black corresponding to the highest intensity). To illustrate the procedure of identifying the phase formation sequence, we will discuss the second *in situ* XRD scan, i.e. the scan of the sample with 33 at.%Ge. From room temperature on, three peaks are clearly visible, indicating the presence of a crystalline phase in the as-deposited film. All these peaks can be attributed to Ni₂Ge: the peak at 2 θ = 41.3° can be indexed as Ni₂Ge (103), the intense peak around 2 θ = 44° is a superposition of Ni₂Ge (031) and Ni₂Ge (211) and the third peak around 2 θ = 47.5° can be indexed as a superposition of Ni₂Ge (020) and Ni₂Ge (113). Around 550 °C, a phase transformation occurs as indicated by the disappearance of the three Ni₂Ge peaks and the emergence of three new peaks. The new features at 2 θ = 43.8° and 51° can be identified as β -Ni₃Ge (111) and β -Ni₃Ge (200) respectively. The third peak around 2 θ = 45° however is not that straightforward to identify, since it can be indexed as either ϵ -Ni₅Ge₃(102), Ni₃Ge₂(102) or Ni₁₉Ge₁₂(212). A complementary *in situ* XRD scan (not shown here) in a different 2 θ window revealed a second peak around 2 θ = 30.5° which can similarly be attributed to the (101) peak of either ϵ -Ni₅Ge₃ or Ni₃Ge₂ or to Ni₁₉Ge₁₂(200). Similar identification difficulties of these specific phases, which exist over a broad composition range in the binary Ni–Ge phase diagram (refer to the shaded area in Fig. 1), were encountered in a vast subset of the measured samples. This identification problem can be related to the very closely related crystal structures of these three phases [3,13], which makes it nearly impossible to discern between them based solely on powder XRD

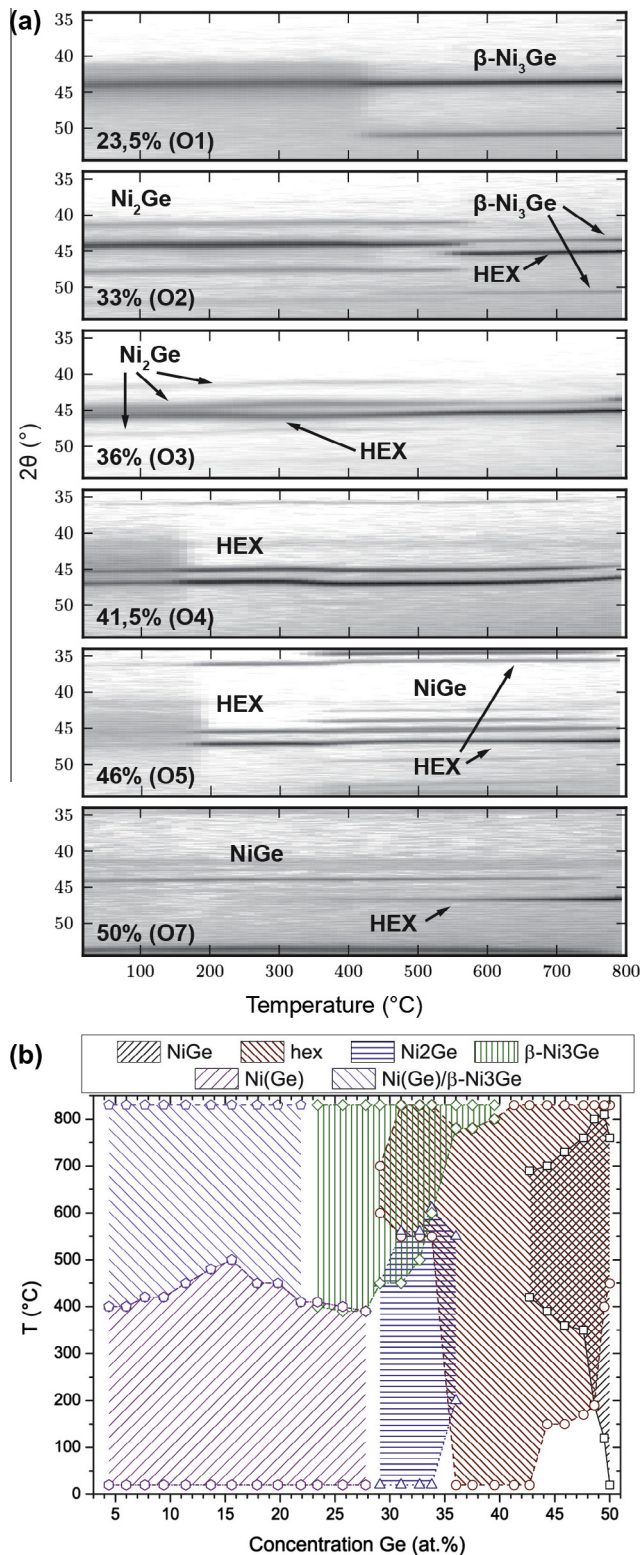


Fig. 2. (a) Selected *in situ* XRD scans representing the different phase formation sequences observed on the oxide samples during an anneal to 800 °C at 3 °C/s. The XRD intensity is plotted as a function of temperature (x-axis) and 2θ (y-axis) using a grayscale map with black representing the highest intensity. (b) Summary of phase formation for Ni (x%Ge) on SiO₂ samples. The shaded areas mark the regions in Ge concentration and temperature where the respective phase was detected with *in situ* XRD.

techniques. Therefore, in our XRD results, we will depict these phases as one single phase labeled 'HEX', referring to their hexagonal crystal structure.

A similar analysis was performed for all samples, i.e. the phases were identified on the *in situ* XRD scans and the temperature where a particular phase appears/disappears was systematically recorded. In total, we could discern between seven different phase formation sequences for the samples on SiO₂:

- O1. Ni (4–28%Ge) → β-Ni₃Ge/Ni (Ge)
- O2. Ni (29–34%Ge) → Ni₂Ge → HEX + β-Ni₃Ge
- O3. Ni (36%Ge) → HEX → Ni₂Ge + HEX → HEX
- O4. Ni (37.5–41.5%Ge) → HEX
- O5. Ni (42.5–47.5%Ge) → HEX → HEX + NiGe → HEX
- O6. Ni (48.5%Ge) → HEX + NiGe
- O7. Ni (49–50%Ge) → NiGe → NiGe + HEX

Representative *in situ* XRD measurements for these cases are shown in Fig. 2(a) (a measurement representing case O6 has been left out since it is the same as for case O7, apart from the as-deposited NiGe phase). For most compositions, the phase sequence can be understood by looking at the relevant region in the Ni–Ge equilibrium phase diagram. However, for samples with a Ge concentration between 36 and 48 at.% (cases O3 through O6), the HEX phase which should only be stable above 300 °C is surprisingly observed to crystallize below 300 °C. For samples with a Ge concentration between 36 at.% and 43 at.%, the HEX phase is even present as-deposited (see Fig. 2), indicating that this phase was formed during co-deposition of Ni and Ge at room temperature. Since one could wonder whether the room temperature HEX phase formation is related to the deposition technique used to co-deposit the films, we investigated Ni–Ge films with Ge concentrations of 40 and 42 at.% that were co-deposited on SiO₂ using Molecular Beam Epitaxy. Also in this case, XRD results (not shown here) clearly revealed the presence of a crystalline HEX phase in the as-deposited state.

In Fig. 2(b), a complete phase formation summary for the samples on SiO₂ is presented. For each identified phase, the respective shaded areas give the region in both temperature and Ge concentration where this particular phase was visible on the *in situ* XRD scans. Within the white regions on this plot, no diffraction peaks were observed, indicating an amorphous mixture.

3.2. Ni (Ge) on reactive Ge (100) substrates

On Ge (100), the phase formation sequence will be influenced by the unlimited supply of Ge from the substrate. We expect the initial crystallization of the film to be similar to the corresponding sample deposited on SiO₂. Subsequently, the reaction proceeds by consuming Ge from the substrate until the whole film has transformed to NiGe, which is expected to be the end phase according to the binary phase diagram.

A similar analysis as for the samples on SiO₂ was performed for the samples on Ge (100). Fig. 3(b) shows a phase formation summary for these samples. Below 28 at.%Ge, the observed phase sequence is qualitatively the same as what has been observed previously for the reaction of a pure Ni film on Ge (100) [6]. Adding over 28 at.%Ge results in a phase sequence with successive phases increasing in Ge content. As for the samples on SiO₂, the phase sequence systematically changes when more Ge is added to the as-deposited film. In summary, we observed the following phase formation sequences for the samples on Ge (100):

- G1. Ni (4–28%Ge) → Ni-rich phase (s) → NiGe
- G2. Ni (29–34%Ge) → Ni₂Ge → HEX + NiGe → NiGe
- G3. Ni (36%Ge) → Ni₂Ge + HEX → HEX + NiGe → NiGe
- G4. Ni (37.5–47.5%Ge) → HEX → NiGe
- G5. Ni (48.5%Ge) → HEX + NiGe → NiGe
- G6. Ni (49–50%Ge) → NiGe

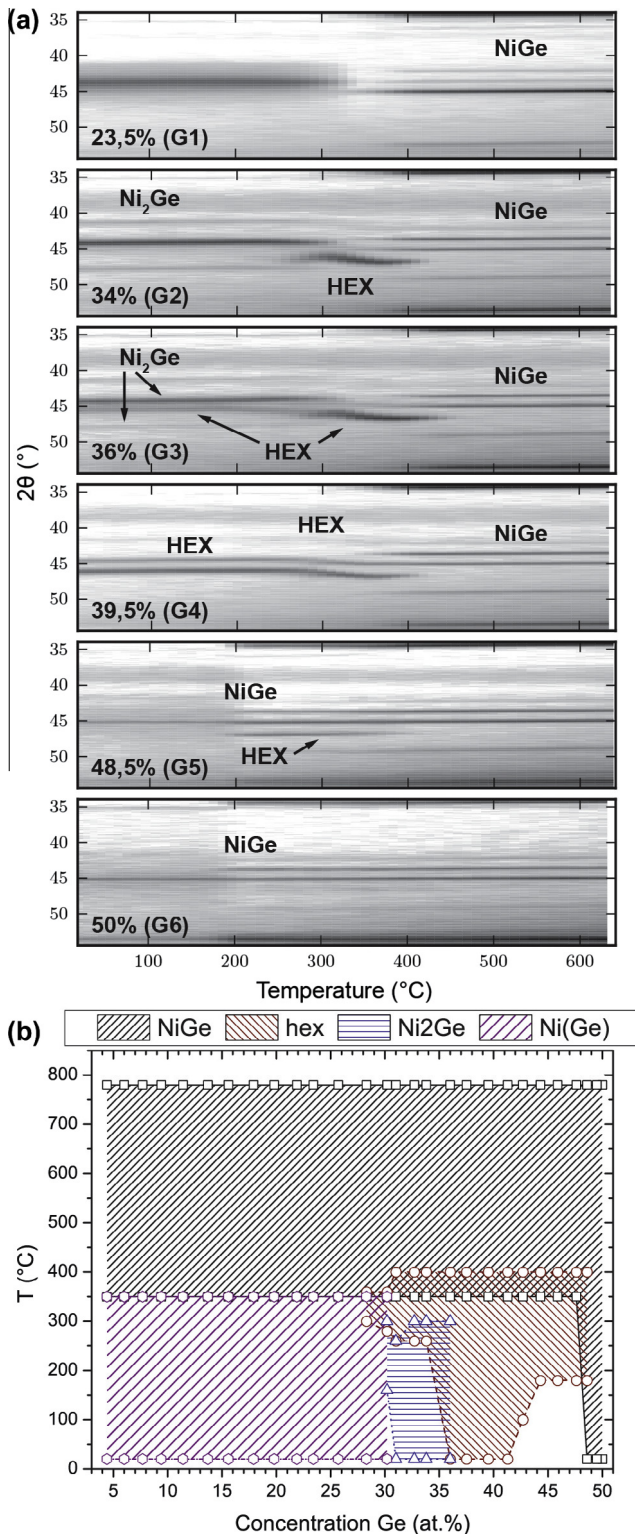


Fig. 3. (a) Selected *in situ* for annealing temperatures above XRD scans representing the different phase formation sequences observed on the Ge (100) samples during an anneal to 650 °C at 3 °C/s. XRD intensity is plotted as a function of temperature (x -axis) and 2θ (y -axis) using a grayscale map with black representing the highest intensity. (b) Summary of phase formation for Ni ($x\%$ Ge) on Ge (100) samples. The shaded areas mark the regions in Ge concentration and temperature where the respective phase was detected with *in situ* XRD.

Again, the selected *in situ* XRD scans visible in Fig. 3(a) are representative for the different cases. Similar to what we have observed for the samples on SiO_2 , the HEX phase is forming below

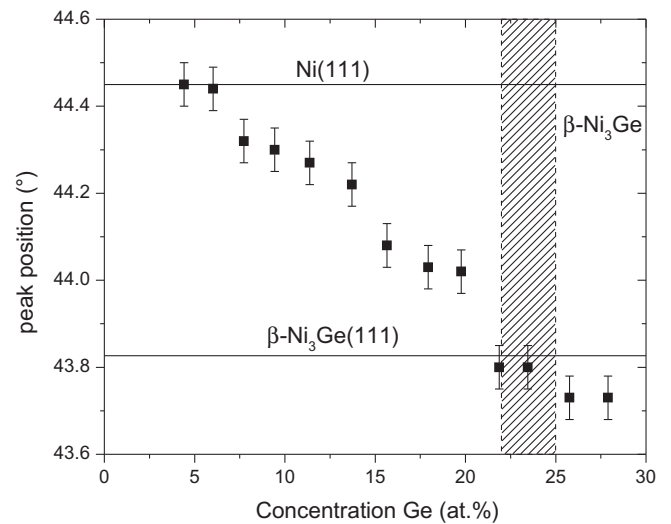


Fig. 4. Evolution of the measured peak position in the as-deposited Ni-Ge layers with Ge concentrations <28 at.% as a function of Germanium concentration. The JCPDS peak positions of Ni (111) and $\beta\text{-Ni}_3\text{Ge}$ (111) are indicated by the horizontal lines. The shaded area depicts the stability range of $\beta\text{-Ni}_3\text{Ge}$ on the binary phase diagram.

300 °C for samples on Ge (100) with a Ge concentration between 36 and 48 at.%. Also in this case, for samples with a Ge concentration between 36 and 42 at.%Ge, the HEX phase is present in the as-deposited film (see Fig. 3)(b), again indicating the formation of this phase during the co-sputtering of Ni and Ge at room temperature.

4. Discussion

In recent years, detailed phase formation studies in the technologically relevant Ni-silicon system have proven to be an important step in understanding the properties and behavior of Ni-silicide phases. For these planar metal thin film on silicon systems, it is generally accepted that phase formation initiates in a thin intermixed amorphous layer which forms at the metal-silicon interface during deposition. Consequently, we can assume that a similar process is true for metal-germanium systems. In this context, the amorphous Ni-Ge layers studied in this work can be considered as simplified models for such an intermixed interface layer.

Based on the phase formation summary plots (Figs. 2(b) and 3(b)), some interesting observations can be made. First, the influence of the unlimited Ge supply from the substrate for the samples on Ge (100) is clearly visible. From Fig. 3(b) it can be seen that NiGe is always the stable end phase for annealing temperatures above ~ 380 °C, which makes sense because an unlimited supply of Ge is expected to push the system towards the most Ge rich germanide phase. In contrast, for the samples on SiO_2 the NiGe phase only forms when the Ge concentration in the intermixed films approaches 50 at.%, which is the Ge content expected in stoichiometric NiGe.

For samples in the ranges O1 and G1, a single (broad) diffraction peak is visible around $2\theta = 44^\circ$ in the *in situ* XRD spectra from the as-deposited state on. This peak could be identified as either Ni (111) ($2\theta = 44.49^\circ$) or $\beta\text{-Ni}_3\text{Ge}$ (111) ($2\theta = 43.83^\circ$) due to the closely related crystal structures of Ni and $\beta\text{-Ni}_3\text{Ge}$. In fact, $\beta\text{-Ni}_3\text{Ge}$ is essentially the Ni structure with site-ordered Ni atoms substituted with Ge, yielding a slightly larger lattice constant [5]. Fig. 4 shows this peak's measured position in the as-deposited layers as a function of Ge concentration and clearly reveals a gradual shift from Ni (111) to $\beta\text{-Ni}_3\text{Ge}$ (111) when approaching the stable composition range of the latter phase (according to the Ni-Ge phase

diagram in Fig. 1, depicted by the shaded area in Fig. 4). The presence of this peak suggests that there is already some local structure in the co-deposited Ni–Ge films that probably resembles the crystal structure of Ni and β -Ni₃Ge. The positional shift of the peak could then be caused by a solid-solution behavior of this intermixed phase, meaning that the addition of extra Ge in the co-deposited film drives the local structure from Ni to β -Ni₃Ge as the extra Ge gets systematically incorporated. Around 400 °C, for the samples on SiO₂, a sudden improvement of the crystalline quality of this solid-solution phase is clearly evidenced by the intensity increase and narrowing of the peak around 44° and the emergence of a second peak around 51°. For the samples on Ge (100), the intermixed phase transforms to NiGe at a similar temperature due to the unlimited supply of Ge from the substrate.

The crystallinity of the as-deposited films is even more pronounced for samples in the ranges O2 and G2. Here, multiple sharp diffraction peaks belonging to Ni₂Ge were found to be present in the co-deposited films, suggesting the presence of fully crystalline Ni₂Ge grains in these films (see Figs. 2(a) and 3(a)).

The most interesting composition region comprises the ranges O3 through O6 for the samples on oxide and G3 through G5 for the samples on Ge (100), i.e. the compositional range between 36 and 48 at.%Ge. *In situ* XRD revealed two clear diffraction peaks around 45° and 46.5° for temperatures below 300 °C. For Ge concentrations in the range [36, 42] at.% these peaks are present from the as-deposited state on, again indicating the existence of crystalline grains in these co-deposited films. According to the Ni–Ge phase diagram (Fig. 1), four phases are expected to be stable below 300 °C and thus could explain the presence of these diffraction peaks: β -Ni₃Ge, Ni₂Ge, NiGe and the low-temperature ϵ' -Ni₅Ge₃ phase. While neither of the two peaks can be attributed to one of the first three phases, they could be explained by superpositions of the ϵ' -Ni₅Ge₃ (203)/(313) and (602)/(331) peaks. Nonetheless, according to the JCPDS data, a lot more diffraction peaks are expected for this ϵ' -Ni₅Ge₃ phase in the measured 2θ window, especially since we do not expect any texture effects on the SiO₂ substrates. Unexpectedly, further investigation of this *in situ* XRD data below 300 °C revealed a perfect agreement with the diffraction pattern that is expected for the high-temperature, hexagonal ϵ -Ni₅Ge₃ and Ni₃Ge₂ phases (which we treat together as a single 'HEX' phase as mentioned above), i.e. all expected diffraction peaks were present and no extra peaks were found. This was confirmed with *in situ* XRD scans taken in a complementary 2θ window (data not shown here). Since this HEX phase is not expected to be stable below 300 °C according to the Ni–Ge phase diagram (Fig. 1), these findings suggest the presence of metastable HEX grains in these films below 300 °C.

To confirm the presence of these metastable HEX grains, we performed a high-resolution TEM (HR-TEM) measurement combined with electron diffraction on a sample in the range G4 that was quenched at 120 °C. The results are shown in Fig. 5. The HR-TEM image (Fig. 5(a)) shows a crystalline Ni–Ge film on top of the Ge (100) substrate. An electron diffraction pattern collected in the indicated area is shown in Fig. 5(b), confirming a hexagonal crystal structure. Fig. 5(c) shows a simulated diffraction pattern of this HEX phase using the lattice parameters known from literature (spacegroup P6₃/mmc; $a = b \approx 3.9$ Å; $c = 5.078$ Å; $\alpha = \beta = 90^\circ$; $\gamma = 120^\circ$) [3,13]. A good agreement between the measured and simulated patterns can be observed, reinforcing the suggested presence of metastable HEX grains in the co-deposited films. Additional HR-TEM results (not shown here) confirm the presence of this phase throughout virtually the whole Ni–Ge layer. From this observation, it can be assumed that the aforementioned two diffraction peaks around 45° and 46.5° that were visible in *in situ* XRD scans of samples in the range [36, 48] at.%Ge are caused by HEX (102) and (110) diffractions respectively. The *in situ* XRD

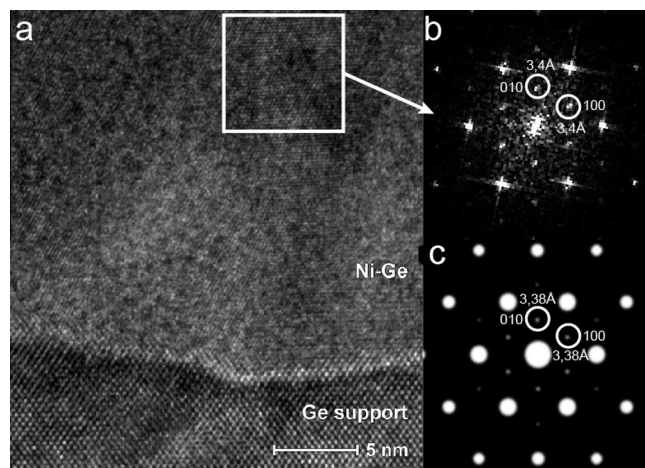


Fig. 5. (a) HR-TEM image of a Ni (38%Ge) film on Ge (100) quenched at 120 °C. (b) Electron diffraction pattern recorded within the indicated area. A similar diffraction pattern could be observed virtually everywhere throughout the film. (c) Simulated diffraction pattern for the HEX crystal structure as known from literature (spacegroup P6₃/mmc; $a \approx 3.9$ Å; $c \approx 5.078$ Å).

results thus suggest the metastable occurrence of this HEX phase in the wide compositional range between 36 and 48 at.%Ge. Between 36 and 42 at.%Ge, the crystallization of this phase occurs already during deposition.

Next, we focus on the influence of Ge content on the HEX crystal structure. In the metastable region discussed above, a systematic shift of the (102) and (110) peaks towards higher diffraction angles over a range of $\sim 2^\circ$ could be observed in the *in situ* XRD measurements, both for the samples on SiO₂ as for those on Ge (100). From the positions of these two diffraction peaks (as determined from the *in situ* XRD scans at 200 °C), the lattice parameters a and c of the HEX phase can be calculated using Bragg's law and the relation between d -spacing and lattice constants for a hexagonal crystal structure. The influence of the Ge content in the co-deposited Ni–Ge film on the lattice of the metastable HEX phase is presented in Fig. 6, which shows a systematic contraction of the lattice with increasing Ge content. A similar behavior has been observed previously by Ellner et al. in the bulk system [3] and by Jensen et al. in multilayer Ni–Ge films [8]. Since the structure of this HEX phase

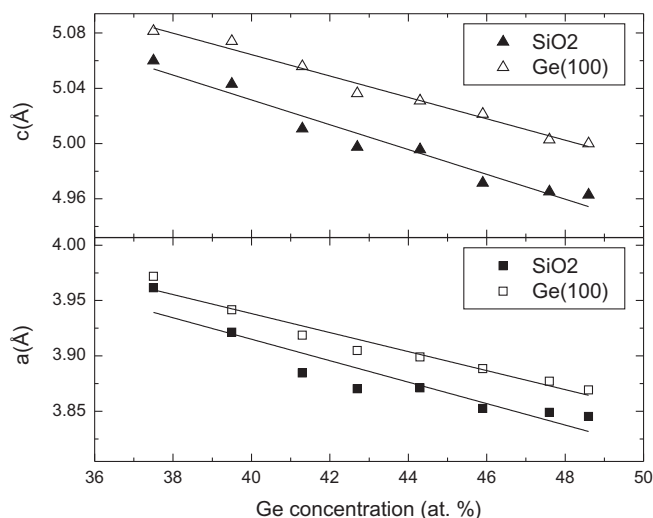


Fig. 6. Influence of the Ge content in the co-deposited film on the lattice parameters a and c of the metastable HEX crystal structure, both on SiO₂ and Ge (100). The parameters were calculated from the HEX (102) and (110) peak positions as extracted from the *in situ* XRD scans at 200 °C.

can be regarded as intermediate between the two prototype structures of Ni_2In and NiAs , the monotonic decrease of the lattice parameters can be explained by a systematic incorporation of extra vacancies on the Ni sites as more Ge is added to the mixture [13,8]. The observation of this metastable HEX phase and its behavior as a function of Ge content is very similar to what has been observed recently for the θ -phase in the Ni–Si system [11].

Finally, the behavior of this HEX phase with increasing Ge content also explains the continuous shift of the HEX (102) and (110) diffraction peaks between 300 and 400 °C that was observed for various samples on Ge (100) (see Fig. 3(a)). First, the HEX phase nucleates with a certain composition out of the co-deposited film as mentioned above. Then, when the thermal budget supplied to the sample is high enough to allow for a reaction between the substrate and the film (in this case around 300 °C), additional Ge from the substrate gets incorporated into the HEX phase, inducing a lattice contraction and a corresponding peak shift towards higher 2θ diffraction angles. Additional TEM measurements on a sample with 38 at.%Ge quenched after the peak shift (not shown here) confirmed the presence of HEX grains with similar electron diffraction patterns than those observed for HEX grains at 120 °C. This confirms our assumption that this peak shift is not due to a phase transformation but to a monotonic compositional change of this HEX phase as explained above.

5. Conclusions

A detailed study of phase formation in co-deposited Ni–Ge thin films with Ge concentrations ranging from 0 to 50 at.% on both SiO_2 and Ge (100) substrates was carried out using *in situ* XRD and TEM. A combinatorial sputter deposition technique allowed for the deposition of all investigated samples in a single run. Crystalline phases were found to nucleate already during co-deposition of the films at room temperature for almost the full composition

range under investigation. Between 36 and 48 at.%Ge, a metastable high-temperature hexagonal germanide (ϵ - Ni_5Ge_3 or Ni_3Ge_2 in the binary phase diagram, but treated as a single ‘HEX’ phase in this work) was found to be present below its minimum stability temperature of 300 °C, adapting its crystal structure to the available amount of Ge. Between 36 and 42 at.%Ge, this HEX phase was present already in the as-deposited state.

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